### Potential Quiz Questions

## **<u>QUIZ 1</u>** (Calibration, standards, strain gages, tensile testing, c/w and annealing)

#### Calibration, Standards (refer to the ME Laboratory booklet)

- In order to have confidence in measurements, the measuring device should be: a) less than 10 years old; b) calibrated and certified; c) be guaranteed by the manufacturer to be accurate
- Calibration is the process of estimating a measuring devices uncertainty (error) compared to known (true) value. True/false
- Standardized tests: a) help assure validity of testing; b) help communication regarding test procedures; c) helps prevent "reinventing the wheel"; d) allows for availability of standard test specimens; e) all of the above.

#### **Strain Gages**

- A Wheatstone bridge circuit is used in the strain indicator because: a) it is unaffected by temperature changes; b) a small change in resistance will produce a measurable voltage changed; c) Wheatstone bridge circuits can measure very small electrical currents.
- What is the characteristic of the gage that relates change in resistance to strain: a) gage factor; b) gage number; c) strain number; d) R to E ratio

### **Tensile Testing**

Cold rolling AISI 1045 steel decreases the strength (compared to hot rolled): true, false

A "reasonable" strain in an aluminum test specimen near its yield strength would be:

a) 0.0048 in b) 0.0048 in/in c) 48,000 psi d) 48,000 ksi

In a standard tensile test, the ultimate tensile strength is equal to: (define)

Young's Modulus (Modulus of Elasticity) of aluminum is: a) less than steel, b) about the same as steel, c) greater than steel

Which of the following properties cannot be obtained from a stress-strain curve: a. Young's modulus; b. Poisson's ratio; c. yield strength; d. tensile strength

# **Cold Working and Annealing of Copper**

- The "recovery" process that occurs during the annealing: a. decreases the number of dislocations; b. increases the number of dislocations; c. has no effect on dislocations
- Annealing strain hardened copper causes it to: a. get harder; b. get softer; c. does not change hardness
- Strain hardening causes dislocation density to: a. increase significantly (creates dislocations); b. decrease significantly (annihilate dislocations); c. not change significantly
- The rate of recrystallization and grain growth during annealing: a. increases as temperature increases; b. decreases as temperature increases; c. is not affected by temperature

# **<u>OUIZ 2:</u>** (HT of alum, Jominy end quench and HT of steel)

### **Precipitation Heat Treating of Aluminum**

Are all aluminum alloys heat treatable (i.e. can they all be precipitation heat treated)?

- Precipitation hardening of 2024 aluminum results in finely dispersed precipitates that: a. facilitate the motion of dislocations; b. impede the motion of dislocations; c. have no effect on the motion of dislocations.
- After quickly quenching 2024 aluminum alloy from solution heat treating temperature (~1000°F) to 35°F, the alloy will: a. be softer compared to pre-heating hardness (as received T351 condition); b. be harder compared to pre-heating hardness (as received T351 condition); c. have precipitates form; d. both b and c

The first step in precipitation heat treating of alloys such as 2024 aluminum is: (name)

- Engineers need to be cautious when using precipitation hardened alloys at elevated temperatures because: a. they are more susceptible to creep; b. they can overage and hence become weaker; c. they have low melting points
- For this lab, we started with 2024-T351. What does the 2024 refer to specifically? What does the T351 refer to specifically? 2024: T351:

Briefly describe what overaging is.

Describe the microstructure of 2024 aluminum alloy during the following: -during solution heat treat -immediately after quenching from solution heat treat -after precipitation heat treating

## Jominy End-Quench and Heat Treating of Steel

Tempering of martensite is typically carried out above the eutectoid temperature: true, false

- AISI 4140 steel nominally contains: a) 40% carbon, by weight; b) 0.40% carbon, by weight; c) 1.40% carbon, by weight
- Rapidly quenching AISI 4140 steel from austenite (above about 1600°F): a. causes it to get softer compared to "hot rolled"; b. causes it to get harder compared to "hot rolled"; c. causes precipitates to form
- AISI 4140 steel is more hardenable than AISI 1040 steel because: a. the chromium and molybdenum in 4140 steel inhibit dislocation motion; b. the chromium and molybdenum in 4140 steel inhibit carbon diffusion; c. the chromium and molybdenum in 4140 steel increase thermal conductivity
- AISI 1040 steel will be harder than AISI 1018 steel after they have been austenitized and quenched primarily because: a. there is more carbon in AISI 1040; b. AISI 1040 is more hardenable; c. AISI 1040 quenches more quickly than AISI 1018; d. none of the above.
- Slow cooling AISI 1040 steel after austenitizing typically results in: a. a mixture of proeutectoid ferrite and pearlite; b. a mixture of proeutectoid ferrite and bainite; c. martensite
- AISI 1018 steel may be precipitation hardened: true false

### **<u>OUIZ 3</u>** (microscopy, welding, Charpy Impact, Residual Stress)

#### Microscopy and welding:

- Welding a highly hardenable steel will likely result in martensite being formed in the weld region. True/False
- Welding a heat treatable aluminum alloy (meaning it can be precipitation hardened) will result in the weld and heat affected zone being harder than the base metal. True/False
- A heat treatable aluminum alloy (meaning it can be precipitation hardened, for example 2024-T351), when welded, may result in microstructural changes in the heat affected zone that would likely a) be softer; b) be harder; c) be same hardness...than the base metal.

- Pearlite, spheroidite, bainite, and tempered martensite are all composed of two phases:  $\alpha$ -ferrite and cementite. True/False
- Martensite is a metastable phase composed of  $\alpha$ -ferrite and cementite. True/False
- The properties of a metal are affected by: a. composition of each phase; b. phases that are present; c. phase morphology (morphology includes shape, size and distribution of phases); d. all of the above
- The ability of a microscope to distinguish between two closely spaced features is referred to as: a. resolution; b. magnification; c. neither a nor b
- An electron microscope can have far better resolution than a light microscope because: a. Electrons are larger than photons; b. Wave length of electron beams can be much shorter than optical light waves; c. Electrons may have lower energy than photons

The microstructure of a metal may be identified by using a light microscope. True/False

The crystal structure of a metal may be identified by using a light microscope. True/False

- The eutectoid composition of plain carbon steel is about 0.8wt%C. If the microstructure of a plain carbon steel is about 25% proeutectoid ferrite and 75% pearlite, its carbon content is: a. about 0.2wt%C; b. about 0.4wt%C; c. about 0.6wt%C; d. about 0.8wt%C.
- An SDS (Safety Data Sheet), formerly MSDS (Materials Safety Data Sheet): a. contains potential health hazards of the chemical; b. contains the manufacturers name; c. must be available for all employees; d. all of the above

### **Charpy Impact Testing**

- The reason the aluminum alloy did not show a ductile to brittle transition was because: a. it has low density; b. the FCC crystal structure allows dislocations to freely move at low temperature; c. the strain energy produced by precipitation hardening improves toughness
- The reason many metals exhibit a ductile to brittle transition is because: a) most metals develop a brittle microstructure at low temperatures; b) diffusion decreases with temperature, and since effectively dislocation motion can be considered a form of diffusion, their ability to move decreases at low temperatures; c) low temperatures increase the cold rolling characteristics of a metal.

The reason dislocations are able to move through metals with FCC crystal structures easily even at very low temperature is because:

- a) FCC has the highest possible planar atomic density
- b) FCC has the highest possible linear atomic density

- c) Both a and b
- Increasing carbon content in plain carbon steel: a. decreases the ductile to brittle transition temperature, b. increases the ductile to brittle transition temperature, c. has little effect on the ductile to brittle transition temperature
- Cold worked steel compared to hot rolled will be: a. tougher, b. less tough, c. no difference in toughness
- Metals composed of what crystal structure do not exhibit ductile to brittle transition: a) FCC; b) BCC; c) BCT

### **Residual Stress**

- Generally, compressive residual stresses on the surface: a. are usually desirable; b. are usually undesirable; c. can not determine if they are desirable or not.
- Residual stresses may be caused by: a. phase changes; b. plastic deformation; c. manufacturing processes; d. all of the above
- At times, manufacturing processes are used to intentionally introduce residual stresses into a part. True/False
- At times, thermal treatments are used to intentionally reduce residual stresses True/False
- Welding can produce residual stresses exceeding 75% of the yield strength of the material True/False